## Metallation Studies with Pyrimidines

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2-Alkylamino-4,6-diphenylpyrimidines are acylated only at nitrogen after treatment with lithium di-isopropylamide (LDA). 4,6-Diaryl-1-benzylpyrimidin-2(1H)-ones can be acylated and alkylated at the $\alpha-\mathrm{CH}_{2}$ group. 1-Methyl-4,6-di-p-tolylpyrimidin-2(1H)-one forms a carbanion which undergoes dimerisation.

As part of a larger study of heterocycle stabilised carbanions, ${ }^{1}$ we have investigated pyrimidines: (a) 2 -alkylamino-4,6-diphenylpyrimidines to attempt to generate carbanions of type (9), and (b) 1 -substituted 4,6-diphenylpyrimidin- $2(1 \mathrm{H})$-ones to examine carbanions of type (18).

(1)

(2)

(3) $R=H, \quad R^{\prime}=M e$
(4) $R=H, R^{\prime}=E t$
(5) $R=H, R^{\prime}=\mathrm{Pr}^{n}$
(6) $R=P h, R^{\prime}=H$
(7) $R=M e R^{\prime}=H$

(8)


(9)
(10) $R^{\prime}=M e$
(11) $\mathrm{R}^{\prime}=\mathrm{Et}$
(12) $\mathrm{R}^{\prime}=\mathrm{Pr}^{\boldsymbol{n}}$
Scheme 1

2-Alkylaminopyrimidines.-4,6-Diphenylpyrimidin$2(1 \mathrm{H})$-one (1) is converted by phosphoryl chloride into 2 -chloro-4,6-diphenylpyrimidine (2). ${ }^{2}$ This reacts readily with primary and secondary amines (ethylamine, npropylamine, n -butylamine, and N -methylaniline) to give the corresponding 2 -alkylaminopyrimidines (3)-(6) (Table 1). The analogous 2-dimethylamino-derivative

(13) $\mathrm{Ar}=\mathrm{Ph}$
a; $R=P h$
(16) $\mathrm{Ar}=\mathrm{Ph}$
(14) $\mathrm{Ar}=p-\mathrm{Tol}$
b; $R=H$
(17) $\mathrm{Ar}=\mathrm{p}-\mathrm{Tol}$



(20) $A r=P h$
(18) $\mathrm{Ar}=\mathrm{Ph}$
(21) $\mathrm{Ar}=p-\mathrm{Tol}$
(19) $\mathrm{Ar}=p-\mathrm{Tol}$

| E | E |
| :---: | :---: |
| $a ; H$ | $f$; Eto |
| b; D | g ; Ph |
| c; Ac | $h ; p-$ |
| d; Bz | i : Me |
| e; $\mathrm{p}-\mathrm{MeC}_{6} \mathrm{H}_{4} \mathrm{CO}$ | j; Et |

Scheme 2
(7) was obtained by ring synthesis from dibenzoylmethane and NN -dimethylguanidine. Structures of all the compounds were confirmed by their spectral data (Table 2).
Treatment of the 2 -alkylamino-4,6-diphenylpyrimidines (3)-(5) with lithium di-isopropylamide (LDA) gave the mono-anion (8), but not the dianion (9) : subsequent treatment with $p$-toluoyl chloride yielded amides (10)-(12) (Table 3). Structures were confirmed by their spectral data (Table 4): in particular the low $v(\mathrm{C}=\mathrm{O})$ excludes the possibility that the acyl group is attached to a cyclic nitrogen atom. Use of excess LDA mainly gave the known ${ }^{3} N N$-di-isopropyl- $p$-toluamide.
4,6-Diaryl-1-benzylpyrimidin-2(1H)-ones.-The diaroylmethanes (13) and (14) reacted with benzylurea (15a) to give the 4,6-diaryl-1-benzylpyrimidin-2(1H)-

Table 1
Preparation of 2-alkylamino-4,6-diphenylpyrimidines (3)-(7)

| No. | R | R' | M.p. $\left({ }^{\circ} \mathrm{C}\right)$ | Yield (\%) | Crystal form | Cryst. solvent | Found (\%) |  |  | Required (\%) |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  |  |  |  |  | C | H | N | C | H | N |
| (3) | H | Me | 66-68 | 49 | Needles | EtOH | 78.2 | 6.1 | 15.1 | 78.5 | 6.2 | 15.3 |
| (4) | H | Et | 96-99 | 51 | Prisms | EtOH | 78.7 | 6.7 | 14.5 | 78.9 | 6.6 | 14.5 |
| (5) | H | $\mathrm{Pr}^{\mathrm{n}}$ | 58-61 | 58 | Plates | EtOH | 78.8 | 6.8 | 13.7 | 79.2 | 7.0 | 13.8 |
| (6) | Ph | H | 147-147.5 | 91 | Fine needles | 95\% EtOH | 81.8 | 5.8 | 12.4 | 81.9 | 5.7 | 12.5 |
| (7) | Me | H | 131.5-132.5 | 30 | Prisms | EtOH | 78.5 | 6.0 | 15.2 | 78.5 | 6.2 | 15.3 |

Table 2
I.r. ${ }^{a}$ and ${ }^{1} \mathrm{H}$ n.m.r. $(\delta, 60 \mathrm{MHz})^{b}$ spectral data of 2 -alkylamino-4,6-diphenylpyrimidines (3)-(7)

| No. (3) | R | R' | $\begin{aligned} & \text { I.r. } \\ & \text { NH } \\ & \left(\mathrm{cm}^{-1}\right) \end{aligned}$ | N.m.r. |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  |  | $=-\mathrm{CH}_{2}-\mathrm{R}^{\prime}$ |  |  |  | Aromatic |
| (3) | H | Me | 3280 | $\begin{gathered} 5.35 \\ (1 \mathrm{H}, \mathrm{bm}) \end{gathered}$ | $\begin{gathered} 3.65 \\ (2 \mathrm{H}, \mathrm{qn}) \end{gathered}$ | $\begin{gathered} 1.33 \\ (3 \mathrm{H}, \mathrm{t}) \end{gathered}$ |  | $\begin{gathered} 7.8 \\ (11 \mathrm{H}, \mathrm{~m}) \end{gathered}$ |
| (4) | H | Et | 3250 | $\begin{aligned} & 5.4 \\ & (1 \mathrm{H}, \mathrm{bm}) \end{aligned}$ | $\begin{gathered} 3.5 \\ (2 \mathrm{H}, \mathrm{q}) \end{gathered}$ | $\begin{aligned} & 1.5 \\ & (2 \mathrm{H}, \mathrm{~m}) \end{aligned}$ | $\left(\begin{array}{l} 1.0 \\ \mathrm{H}, \mathrm{t}) \end{array}\right.$ | $\begin{gathered} 7.8 \\ (11 \mathrm{H}, \mathrm{~m}) \end{gathered}$ |
| (5) | H | $\mathrm{Pr}^{\mathrm{n}}$ | 3280 | $\begin{gathered} 5.9 \\ (1 \mathrm{H}, \mathrm{t}) \end{gathered}$ | centred at 3.35 <br> ( $2 \mathrm{H}, \mathrm{bm}$ ) | $\begin{gathered} 0.9 \\ (3 \mathrm{H}, \mathrm{~d}) \end{gathered}$ | $\begin{gathered} 1.4 \\ (4 \mathrm{H}, \mathrm{~m}) \end{gathered}$ | $\begin{gathered} 7.8 \\ (11 \mathrm{H}, \mathrm{~m}) \end{gathered}$ |
| (6) | Ph | H |  |  | $\begin{gathered} 3.68 \\ (3 \mathrm{H}, \mathrm{~m}) \end{gathered}$ |  |  | $\begin{aligned} & 7.12-7.68 \\ & (12 \mathrm{H}, \mathrm{~m}) \\ & 7.9-8.20 \\ & (4 \mathrm{H} . \mathrm{m}) \end{aligned}$ |
| (7) | Me | H |  |  | $\begin{array}{r} 3.34 \\ (6 \mathrm{H}, \mathrm{~s}) \end{array}$ |  |  | $\begin{aligned} & 7.36-7.66 \\ & (7 \mathrm{H}, \mathrm{~m}) \\ & 8.17-8.2 \\ & (4 \mathrm{H}, \mathrm{~m}) \end{aligned}$ |

${ }^{a} \operatorname{In} \mathrm{CHBr}_{3} . \quad{ }^{b} \mathrm{In} \mathrm{CDCl}_{3} . \quad \mathrm{s}=$ singlet, $\mathrm{bm}=$ broad multiplet, $\mathrm{t}=$ triplet, $\mathrm{q}=$ quartet, $\mathrm{q} \mathrm{n}=$ quintet, $\mathrm{m}=$ multiplet.
Table 3
Preparation of $N$-alkyl- $N$-(4,6-diphenylpyrimidinyl)- $p$-toluamides (10)-(12)

| No.$(10)$ | $\mathrm{R}^{\prime}$ | M.p. $\left({ }^{\circ} \mathrm{C}\right) \quad \begin{gathered}\text { Yield } \\ (\%)\end{gathered}$ |  | Cryst. form | Cryst. solvent | Found (\%) |  |  | Required (\%) |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  |  | C |  | H | N | C | H | N |
|  | Me | 128-132 | 60 |  | Needles | Ether-light petroleum <br> (b.p. $60-80^{\circ} \mathrm{C}$ ) | 78.9 | 5.6 | 10.6 | 79.4 | 5.9 | 10.7 |
| (11) | Et | 135-138 | 55 | Prisms | EtOH | 79.4 | 6.4 | 10.3 | 79.6 | 6.2 | 10.3 |
| (12) | $\mathrm{Pr}^{\mathbf{n}}$ | 91-94 | 37 | Needles | 95\% EtOH | 79.5 | 6.5 | 9.9 | 79.8 | 6.5 | 10.0 |

Table 4
I.r. ${ }^{a}$ and ${ }^{1} \mathrm{H}$ n.m.r. $(\delta, 60 \mathrm{MHz})^{b}$ spectral data of $N$-alkyl- $N$-(4,6-diphenylpyrimidinyl)-p-toluamides (10)-(12)

| No. | $\mathbf{R}^{\prime}$ | $\begin{gathered} \text { I.r. } \\ \left(\mathrm{cm}^{-1}\right) \\ >=0 \end{gathered}$ | N.m.r. |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  |  |  |  |  |  |
|  |  |  | $p$-Toluoyl | $\mathrm{N}-\mathrm{CH}_{2}$ | - $\mathrm{R}^{\prime}$ |  | Aromatic |
| (10) | Me | 1655 | 2.27 | 4.42 | 1.4 |  | 7.4 |
|  |  |  | (3 H, s) | ( $2 \mathrm{H}, \mathrm{m}$ ) | ( $3 \mathrm{H}, \mathrm{t}$ ) |  | (15 H, m) |
| (11) | Et | 1650 | 2.25 | 4.35 | centred at | 1.05 | 7.5 |
|  |  |  | ( $3 \mathrm{H}, \mathrm{s}$ ) | ( $2 \mathrm{H}, \mathrm{t}$ ) | $\begin{aligned} & 1.9 \\ & (2 \mathrm{H} . \mathrm{m}) \end{aligned}$ | ( $3 \mathrm{H}, \mathrm{t}$ ) | $(15 \mathrm{H}, \mathrm{m})$ |
| (12) | $\mathrm{Pr}^{\mathbf{n}}$ | 1650 | 2.28 | 4.37 | $\left(\begin{array}{c}2 \mathrm{H}, \mathrm{m} \\ 1.0\end{array}\right.$ | 1.65 | 7.5 |
|  |  |  | (3 H, s) | ( $2 \mathrm{H}, \mathrm{t}$ ) | ( $3 \mathrm{H}, \mathrm{t}$ ) | (4 H, m) | (15 H, m) |

ones (16) and (17) (cf. ref. 4). We have shown previously that 1-benzyl-4,6-diphenyl-2(1H)-pyridone (22) is converted by LDA into the lithio-derivative (23) which reacts with electrophiles to form $\alpha$-substituted products (24). We now find that lithio-derivatives (18) and (19) can be formed similarly; they show an intense blue colouration.

The diphenyl carbanions (18) and (19) react with $\mathrm{D}_{2} \mathrm{O}$ to form the deuteriated pyrimidinones (20b) and (21b). Organolithium (18) adds to a variety of electrophiles to
form products ( $20 \mathrm{c}-\mathrm{j}$ ); methyl and ethyl iodide gave the alkylated derivatives (20i) and (20j); ethyl chloroformate the ester (20f); acetyl, benzoyl, and $p$-toluoyl chloride the ketones (20c), (20d), and (20e); p-tolualdehyde and benzophenone the hydroxy-derivatives (20h) and $(20 \mathrm{~g})$. Organolithium (19) reacted with methyl iodide to give the alkylated derivative (21i). All the compounds were characterised by their spectral data (Table 5).

1-Methyl-4,6-di-p-tolylpyrimidin-2(1H)-one.-Reaction

Table 5
I.r. ${ }^{a}$ and ${ }^{1} \mathrm{H}$ n.m.r. $(\delta, 60 \mathrm{MHz})^{b}$ spectral data of 4,6-diaryl-1-benzylpyrimidin-2-ones (20) and (21)

| No. | Alpha substituent | I.r. $\left(\mathrm{cm}^{-1}\right)$ |  |  | ${ }^{1} \mathrm{H}$ N.m.r. |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  | Pyridone$\mathrm{C}=\mathrm{O}$ | R |  | $5-\mathrm{H}(\mathrm{~s})$ | 1'-H | R |  |  |
|  |  |  | $\bigcirc$ | $\mathrm{C}=\mathrm{O}$ |  |  |  |  | Aromatic (m) |
| (20a) | H | 1645 |  |  | 6.70 | 5.2 (s) |  |  | $7.52-7.05(15 \mathrm{H})$ |
| (20b) | D | 1650 |  |  | 6.70 | 5.2 (s) ${ }^{\text {c }}$ |  |  | $7.05-7.52(15 \mathrm{H})$ |
| (20c) | MeCO | 1640 |  | 1680 | 6.42 | d |  | 2.3 (s) | $6.80-7.82(16 \mathrm{H})$ |
| (20d) | PhCO | 1650 |  | 1700 | 6.80 | 6.65 (s) |  |  | $7.2-8.3(20 \mathrm{H})$ |
| (20e) | $p-\mathrm{MeC}_{6} \mathrm{H}_{4} \mathrm{CO}$ | 1650 |  | 1690 | 6.72 | $d$ |  | 2.28 (s) | $7.0-8.2(20 \mathrm{H})$ |
| (20f) | EtOCO ${ }^{\text {P }}$ | 1650 |  | 1740 | 6.40 | 6.30 (s) | 4.0 (m) | 1.1 (t) | $6.98-7.9$ (15 H) |
| (20g) | $\mathrm{Ph}_{2} \mathrm{COH}$ | 1650 | 3250 |  | 6.16 | 6.0 (s) |  |  | $6.74-7.3(25 \mathrm{H})$ |
| (20h) | $p-\mathrm{MeC}_{6} \mathrm{H}_{4} \mathrm{CHOH}$ | 1650 | 3300 |  | 6.50 | 5.86 (d) ${ }^{e}$ | 5.45 (d) ${ }^{e}$ | 2.24 (s) | $7.0-8.0$ ( 14 H ) |
| (20i) | $\mathrm{Me}^{\text {a }}$ | 1650 |  |  | 6.90 | 5.5 (q) |  | 1.6 (d) | $7.2-7.8(15 \mathrm{H})$ |
| (20j) | Et | 1655 |  |  | 6.50 | 5.9 (m) | 1.2 (m) | 0.7 (t) | $6.9-7.9(15 \mathrm{H})$ |
| (21a) $f$ | H | 1650 |  |  | 6.60 | 5.2 (s) |  |  | $7.9-7.1(13 \mathrm{H})$ |
| (21b) $f$ | D | 1655 |  |  | 6.60 | 5.15 (s) ${ }^{\text {c }}$ |  |  | $7.0-8.0(13 \mathrm{H})$ |
| (21i) $f$ | Me | 1650 |  |  | 6.50 | 5.4 (q) |  | 1.75 (d) | $7.85-7.05$ (13 H) |

${ }^{a}$ In $\mathrm{CHBr}_{3} . \quad{ }^{b}$ In $\mathrm{CDCl}_{3} . \quad \mathrm{d}=$ doublet, $\mathrm{q}=$ quartet, $\mathrm{s}=$ singlet, $\mathrm{t}=$ triplet, $\mathrm{m}=$ multiplet. $\quad$. Integrates for $1 \mathrm{H} . \quad{ }^{d}$ Overlapped in the aromatic region. ${ }^{e} \mathrm{CH}$, doublet $(J 10 \mathrm{~Hz}) .{ }^{f}$ In addition two singlets ( 3 H each) are shown at 2.3 and 2.4 p.p.m.
of $N$-methylurea (15b) with (14) yielded 1 -methylpyrimidinone (25) which formed a deep red colouration with LDA; subsequent addition of electrophiles then merely gave recovered (25). Attempted use of n-butyllithium as base afforded the addition product (30), which shows $v(\mathrm{NH}) 3200 \mathrm{~cm}^{-1}$ and $v(\mathrm{C}=\mathrm{O}) 1650$ (broad).


In the ${ }^{1} \mathrm{H}$ n.m.r., the n-butyl group signals occur at 0.84 $(3 \mathrm{H}, \mathrm{t}), 1.3(4 \mathrm{H}, \mathrm{m})$, and $1.82 \mathrm{p} . \mathrm{p} . \mathrm{m} .(2 \mathrm{H}, \mathrm{m})$. The $C$ methyls resonate as two $3-\mathrm{H}$ singlets near to 2.8 p.p.m. The heterocyclic ring olefinic hydrogens form a 2 Hz split doublet at 4.87 p.p.m. coupled ( $J 2 \mathrm{~Hz}$ ) with the $\mathrm{N}-\mathrm{H}$ proton (broad doublet at 5.74 p.p.m.). These spectral data appear to support a 1,2 -addition to $\mathrm{C}=\mathrm{N}$ to give (30) rather than 1,4 -addition to $\mathrm{C}=\mathrm{C}-\mathrm{C}=\mathrm{N}$ to give an isomer, but we have not rigorously excluded the isomeric structure. Sodium borohydride reduces pyrimidin-2ones to dihydro- and tetrahydro-derivatives, ${ }^{5}$ and this provides an analogy for the formation of (30).

However, reaction of (25) with 2,2,6,6-tetramethyl-piperidyl-lithium (LTMP) followed by methyl iodide formed (28) by a 1,6 -addition of the lithio-derivative (26) to a second molecule of (25) to give anion (27). Similarly addition of ethyl iodide produced (29).

The structures (28) and (29) are based on spectral evidence: both (28) and (29) showed $\nu(\mathrm{C}=\mathrm{O})$ (pyrimidinone) at $1650 \mathrm{~cm}^{-1}$ (broad). In the ${ }^{1} \mathrm{H}$ n.m.r. spectra, the ring proton of the undisturbed pyrimidinone ring in (28) gave a singlet at $\delta 6.56\left(5^{\prime}-\mathrm{H}\right)$ and at $\delta 6.62$ in (29) $\left(5^{\prime}-\mathrm{H}\right)$. The corresponding signal in the pyrimidinone (25) was at $\delta 6.64$. In the modified pyrimidinone ring of (28) and (29), the $5-\mathrm{H}$ signal shifted upfield, resonating as a singlet at $\delta 4.54$ and 4.56 respectively. The prochiral
protons of the bridging methylene group were magnetically non-equivalent and gave an AB system in both compounds. In (28) the A-proton appeared at $\delta 4.75$ and the B-proton at $\delta 5.34$ with $J_{\text {gem }} 12 \mathrm{~Hz}$. Similarly in (29) the A-proton appeared at $\delta 4.76$ and the B-proton at $\delta$ 5.41 with $J_{\text {gem }} 14 \mathrm{~Hz}$. In (28) and (29) the $N$-methyl group ( $1^{\prime \prime \prime}$ protons) gave a singlet at $\delta 1.70$; in (28) the incorporated methyl gave a singlet at $\delta 2.5$ and in (29), the diastereotopic methylene protons of the incorporated


$T=p-$ tolyl

(30)
Scheme 3
ethyl formed two sextets at $\delta 3.10$ and $\delta 3.90$ with $J_{\text {vic }}$ 7 Hz and $J_{\text {gem }} 14 \mathrm{~Hz}$. The methyl group displayed a triplet at $\delta 0.78$ p.p.m. The aromatic protons in both (28) and (29) displayed a multiplet in the range $\delta 8.0-7.0$.

In the ${ }^{13} \mathrm{C}$ n.m.r. spectra, with off-resonance $\mathrm{C}-\mathrm{H}$ information, for both (28) and (29), the undisturbed pyrimidinone carbonyl carbon gave a singlet at $\delta 169.19$ and at 169.23 respectively [cf. 169.31 in (25)] whilst the modified pyrimidinone carbonyl resonated at $\delta 160.27$ and 160.12 . The aromatic carbons displayed a series of

HA-100 ( 100 MHz ) n.m.r. spectrometers, ${ }^{13} \mathrm{C}$ n.m.r. spectra at 25.05 MHz on a Jeol FX-100 Fourier transform spectrophotometer, and high-resolution mass spectra on an AEI MS-9 spectrometer.

4,6-Diphenylpyrimidin-2-one (1).-Dibenzoylmethane (5 $\mathrm{g}, 22.3 \mathrm{mmol}$ ), urea ( $2.0 \mathrm{~g}, 33.4 \mathrm{mmol}$ ), toluene-p-sulphonic acid ( $5.75 \mathrm{~g}, 33.4 \mathrm{mmol}$ ), and glacial HOAc ( 15 ml ) were heated under reflux for 48 h and neutralised with aqueous NaOH ( $12 \%$ ). The product separated; it was collected, washed with water, dried, and crystallised ( EtOH ) (m.p. $\left.237-239{ }^{\circ} \mathrm{C}\right)(4.20 \mathrm{~g}, 15 \%)\left[\mathrm{lit.}^{2},{ }^{2}\right.$ m.p. $237-239^{\circ} \mathrm{C}$ ].

Table 6
${ }^{13} \mathrm{C}$ N.m.r. spectra ${ }^{a}$ of 1,3-disubstituted 1,2,3,6-tetrahydro-6-(1,2-dihydro-2-oxo-4,6-di-p-tolylpyrimidinylmethyl)-4,6-di- $p$-tolylpyrimidin-2(1H)-ones (28) and (29)

| Compd. no. | Carbonyl region (s) |  | Aromatic region Unassigned multiplet | Olefinic region |  | Aliphatic region |  |  |  |  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  | $\overbrace{5-C}$ | $\xrightarrow[5]{ }{ }^{\prime}-\mathrm{C}$ |  | $6 \mathrm{C}$(s) | $\underset{(\mathrm{t})}{\mathbf{1}^{\prime \prime}-\mathrm{C}}$ | $\begin{gathered} 1^{\prime \prime \prime}-\mathrm{C} \\ (\mathrm{q}) \end{gathered}$ | $\underbrace{}_{\substack{\text { Aryl-CH } \\ \text { (q) }}}$ |  |  |  | 3-N-R |  |
|  | ${ }_{2} \mathrm{-CO}$ | $2^{\prime} \mathrm{CO}$ |  | (d) | (d) |  |  |  |  |  |  |  |  |  |  |
| (28) | 169.19 | 160.27 |  | 159.68-126.64 | 105.30 | 103.60 | (28) | 67.15 | 48.53 | 30.35 | 21.48 | 21.38 | 21.28 | 20.91 | 33.08 (q) |  |
| (29) | 169.23 | 160.12 | 159.39-126.55 | 105.20 | 103.84 | (29) | 67.73 | 48.04 | 30.26 | 21.50 | 21.39 | 21.19 | 20.90 | 39.27 (t) | 14.96 (q) |

${ }^{a}$ In $\mathrm{CDCl}_{3}$ with $\mathrm{SiMe}_{4}$ as internal reference.
lines in the range $159.68-126.55$ (Table 6). The 5 and $5^{\prime}$ ring carbons gave doublets at $\delta 105.30$ and $\delta 103.60$ in (28) and at $\delta 105.20$ and $\delta 103.84$ in (29) [cf. 102.44 in (25)]. In the aliphatic region, the bridging methylene carbon gave a triplet at $\delta_{\mathrm{av}} 48$, the 6 -quaternary carbon a singlet at $\delta_{\text {ar. }} 67$, and the $N$-methyl, a quartet at $\delta_{\text {av }}$. 30 (Table 6). The inserted methyl in (28) gave a quartet at $\delta 33.08$ whilst in (29), the inserted ethyl gave a triplet at $\delta 39.27$ and a quartet at 14.96 . The $C$-methyl group gave 4 quartets in the range $\delta 21.50-20.90$ (Table 6).

High-resolution mass spectroscopy showed the molecular ion peak at $m / e 594.29(0.12 \%)$ for (28) and at


(31) $R=M e$
(32) $R=E t$
$608.31(0.05 \%)$ for (29). Both (28) and (29) underwent a retrosynthetic expulsion of the oxopyrimidinylmethyl fragment to give the observed base peaks at $m / e 305.16$ ( $100 \%$ ) ( 31 ) and 319.17 ( $100 \%$ ) (32) respectively.

## EXPERIMENTAL

M.p.s were measured on a Reichert hot-stage melting point apparatus and are uncorrected. I.r. spectra were recorded in $\mathrm{CHBr}_{3}$ on a Perkin-Elmer 297 spectrophotometer, ${ }^{1} \mathrm{H}$ n.m.r. spectra on Perkin-Elmer R12 ( 60 MHz ) and Varian

2-Chloro-4,6-diphenylpyrimidine (2) (47\%), had m.p. $113-113.5^{\circ} \mathrm{C}\left[\right.$ lit., ${ }^{2 a}$ m.p. $\left.115-116{ }^{\circ} \mathrm{C}\right]$.

General Procedure for the Preparation of 2-Amino-4,6diphenylpyrimidines (3)-(6).-2-Chloro-4,6-diphenylpyrimidine (2) and the appropriate amine were heated under reflux in absolute EtOH for 2 h . [For the preparation of (3), dry $\mathrm{EtNH}_{2}$ gas was bubbled through a solution of (2) in absolute EtOH and for (6), the reactants were heated in the absence of absolute EtOH]. Evaporation $\left(100{ }^{\circ} \mathrm{C} / 15\right.$ mmHg ) gave the crude product, which was washed with water and crystallised (see Table 1).

2-Dimethylamino-4,6-diphenylpyrimidine (7).-Dibenzoylmethane ( l g, 4.46 mmol ), $N N$-dimethylguanidine, $\mathrm{HCl}(1 \mathrm{~g}$, $8.16 \mathrm{mmol})$, and $\mathrm{K}_{2} \mathrm{CO}_{3}(0.6 \mathrm{~g}, 4.3 \mathrm{mmol})$ were heated under reflux in EtOH ( 10 ml ) for 2 h then extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ $(50 \mathrm{ml})$ and the extracts washed with $\mathrm{H}_{2} \mathrm{O}(20 \mathrm{ml})$. The dried $\left(\mathrm{MgSO}_{4}\right)$ extracts on evaporation ( $100{ }^{\circ} \mathrm{C} / 15 \mathrm{mmHg}$ ) furnished the product, which was crystallised (see Table 1).

General Procedure for the Preparation of Amides (10)-(12). -To LDA ( 1 mmol ) in dry THF ( 5 ml ) [prepared by adding dropwise n -butyl-lithium in hexane ( 1 mmol ) to di-isopropylamine ( 1 mmol ) at 0 to $-5^{\circ} \mathrm{C}$ under $\mathrm{N}_{2}$ ] cooled to 0 to $-5{ }^{\circ} \mathrm{C}$ was added 2-alkylaminopyrimidine (3)-(5) ( 0.5 mmol ) in dry THF $(2 \mathrm{ml})$. After 0.5 h at 0 to $-5{ }^{\circ} \mathrm{C}, p$-toluoyl chloride ( 1 mmol ) in dry THF ( 2 ml ) was added. Stirring was continued for a further 4 h at $20^{\circ} \mathrm{C}$, water ( 5 ml ) was then added and the solution extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(30 \mathrm{ml})$; the extracts were washed with aqueous $\mathrm{NaHCO}_{3}(10 \%, 10 \mathrm{ml})$ followed by $\mathrm{H}_{2} \mathrm{O}(10 \mathrm{ml})$. The dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ extracts on evaporation ( $100^{\circ} \mathrm{C} / 15 \mathrm{mmHg}$ ) gave the crude product as a yellowish oil which solidified slowly and crystallised from the appropriate solvent (Table 3). Purification of the motherliquor afforded $N N$-di-isopropyl- $p$-toluamide as the byproduct, m.p. $84{ }^{\circ} \mathrm{C}$ [lit., ${ }^{3}$ m.p. $\left.85-86{ }^{\circ} \mathrm{C}\right]$ as colourless plates [toluene-light petroleum (b.p. $40-60^{\circ} \mathrm{C}$ )].

1-Benzyl-4,6-diphenylpyrimidin-2(1H)-one (20a).-Dibenzoylmethane ( $5.0 \mathrm{~g}, 20 \mathrm{mmol}$ ), benzylurea ( $4.8 \mathrm{~g}, 32 \mathrm{mmol}$ ), and toluene- $p$-sulphonic acid ( $7.6 \mathrm{~g}, 40 \mathrm{mmol}$ ) in glacial HOAc ( 8 ml ) were heated at reflux for 36 h . Cooling and treatment with aqueous ( $50 \%$ ) EtOH gave the pyrimidinone
which crystallised from $95 \%$ EtOH as prisms ( $5.0 \mathrm{~g}, 68 \%$ ), m.p. $164-165{ }^{\circ} \mathrm{C}$ (Found: C, 81.3; H, 5.4; N, 8.3. $\mathrm{C}_{23}{ }^{-}$ $\mathrm{H}_{18} \mathrm{~N}_{2} \mathrm{O}$ requires C, 81.6; H,5.3; $\mathrm{N}, 8.3 \%$ ).

1-Benzyl-4,6-di-p-tolylpyrimidin-2(1H)-one (21a).—Di-ptoluoylmethane ( $1.0 \mathrm{~g}, 4 \mathrm{mmol}$ ), benzylurea ( $0.95 \mathrm{~g}, 6 \mathrm{mmol}$ ), and toluene- $p$-sulphonic acid ( $1.4 \mathrm{~g}, 7 \mathrm{mmol}$ ), in HOAc ( 1 ml ) were heated at reflux for 36 h . Work-up as described for (20a) gave the pyrimidinone (2la) ( $0.9 \mathrm{~g}, 62 \%$ ) as needles from $95 \%$ EtOH, m.p. $209-210^{\circ} \mathrm{C}$ (Found: C, $81.9 ; \mathrm{H}$, $6.1 ; \mathrm{N}, 7.6 . \mathrm{C}_{25} \mathrm{H}_{22} \mathrm{~N}_{2} \mathrm{O}$ requires $\left.\mathrm{C}, 82.0 ; \mathrm{H}, 6.0 ; \mathrm{N}, 7.6 \%\right)$.

1-Methyl-4,6-di-p-tolylpyrimidin-2(1H)-one (25).-Di-ptoluoylmethane ( $5.0 \mathrm{~g}, 19 \mathrm{mmol}$ ), methylurea ( $3.1 \mathrm{~g}, 40$ mmol ), and toluene- $p$-sulphonic acid ( $7.2 \mathrm{~g}, 37 \mathrm{mmol}$ ) in HOAc ( 9 ml ) were heated at reflux for 36 h . Work-up as described for (20a) gave the pyrimidinone (25) ( $3.5 \mathrm{~g}, 61 \%$ ) as needles from $95 \% \mathrm{EtOH}$, m.p. $197-199^{\circ} \mathrm{C}$ (Found: C, $78.7 ; \mathrm{H}, 6.3 ; \mathrm{N}, 9.6 . \quad \mathrm{C}_{19} \mathrm{H}_{18} \mathrm{~N}_{2} \mathrm{O}$ requires $\mathrm{C}, 78.6 ; \mathrm{H}, 6.2$; $\mathrm{N}, 9.6 \%) ; \nu_{\text {max. }}\left(\mathrm{CHBr}_{3}\right) 1645 \mathrm{~cm}^{-1}(\mathrm{C}=\mathrm{O}) ; \delta\left(\mathrm{CDCl}_{3}\right) 7.98-$ $7.22(8 \mathrm{H}, \mathrm{m}), 6.70(1 \mathrm{H}, \mathrm{s}), 2.40(3 \mathrm{H}, \mathrm{s}), 2.39(3 \mathrm{H}, \mathrm{s})$, and 3.40 ( $3 \mathrm{H}, \mathrm{s}$ ).

General Procedure for the Lithiation and Alkylation of 4,6-Diaryl-1-benzylpyrimidin-2(1H)-ones.-LDA ( 3.0 mmol ) was prepared by adding dropwise di-isopropylamine $(0.3 \mathrm{~g}, 3.0$ mmol) to n-butyl-lithium in hexane ( $3.1 \mathrm{ml}, 3.0 \mathrm{mmol}$ of $0.96 \mathrm{~m})$ at $-20{ }^{\circ} \mathrm{C}$ under $\mathrm{N}_{2}$. Stirring of the mixture was continued until it became cloudy when dry THF ( 6 ml ) added; the whole was then cooled to $-76^{\circ} \mathrm{C}$ and 4,6 -diaryl-1-benzylpyrimidin-2-one ( 3.0 mmol ) in dry THF ( 20 ml ) was added. After 40 min at $-76{ }^{\circ} \mathrm{C}$, the electrophile ( 3.0 mmol ) in dry THF ( 5 ml ) was added. Stirring was continued for 1 h at $-76{ }^{\circ} \mathrm{C}$ and for a further 10 h at $20^{\circ} \mathrm{C}$. Water ( 1 ml ) was then added and the solvent removed at $40-50{ }^{\circ} \mathrm{C} / 20$ mmHg . The residue in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(70 \mathrm{ml})$ was washed with saturated aqueous $\mathrm{NaCl}(30 \mathrm{ml})$ and water $(30 \mathrm{ml})$ and then dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, and evaporated at $40-50{ }^{\circ} \mathrm{C} / 20 \mathrm{mmHg}$. The product was separated by crystallisation [ $\mathrm{CHCl}_{3}-$ light petroleum (b.p. $60-80^{\circ} \mathrm{C}$ )], or prep. t.l.c. and further recrystallised from the appropriate solvent.

The following compounds were prepared according to the general procedure: 1-( $\alpha$-deuteriobenzyl $)-4,6$-diphenylpyrimi-din- $2(1 \mathrm{H}$ )-one ( 20 b ) $(90 \%$ ), prisms from EtOH, m.p. 160 $161{ }^{\circ} \mathrm{C}$ (Found: $\mathrm{C}, 81.5$; N, 8.2. $\mathrm{C}_{23} \mathrm{H}_{17} \mathrm{DN}_{2} \mathrm{O}$ requires C , 81.4; $\mathrm{N}, 8.2 \%$ ); 1-( $\alpha$-acetylbenzyl) $-4,6$-diphenylpyrimidin$2(1 \mathrm{H})$-one $(20 \mathrm{c})(48 \%)$, plates from EtOH, m.p. $184-185{ }^{\circ} \mathrm{C}$ (Found: $\mathrm{C}, 78.7 ; \mathrm{H}, 5.7$; N, 7.4. $\mathrm{C}_{25} \mathrm{H}_{20} \mathrm{~N}_{2} \mathrm{O}_{2}$ requires C , $78.9 ; \mathrm{H}, 5.3 ; \mathrm{N}, 7.4 \%)$; $1-(\alpha$-benzoylbenzyl $)-4,6$-diphenyl-pyrimidin-2 (1H)-one (20d) (52\%), plates from EtOH, m.p. $168-170^{\circ} \mathrm{C}$ (Found: C, 81.4; H, 5.0; N, 6.3. $\mathrm{C}_{30} \mathrm{H}_{22} \mathrm{~N}_{2} \mathrm{O}_{2}$ requires $\mathrm{C}, 81.4 ; \mathrm{H}, 5.0 ; \mathrm{N}, 6.3 \%)$; 4,6-diphenyl-1-[ $\alpha$-(4-toluoyl)benzyl]pyrimidin- $2(1 \mathrm{H}$ )-one ( 20 e ) ( $54 \%$ ), plates from EtOH, m.p. 140-142 ${ }^{\circ} \mathrm{C}$ (Found: C, 81.2; H, 5.6; N, 6.0. $\mathrm{C}_{31} \mathrm{H}_{24} \mathrm{~N}_{2} \mathrm{O}_{2}$ requires C, $\left.81.6 ; \mathrm{H}, 5.3 ; \mathrm{N}, 6.1 \%\right)$; $1-(\alpha-$ ethoxycarbonylbenzyl)-4,6-diphenylpyrimidin-2(1H)-one (20f) ( $35 \%$ ) (isolated by prep. t.l.c.), prisms from $95 \% \mathrm{EtOH}$, m.p. $228-230{ }^{\circ} \mathrm{C}$ (Found: C, 76.0; H, 6.8; N, 5.3. $\mathrm{C}_{26}{ }^{-}$ $\mathrm{H}_{22} \mathrm{~N}_{2} \mathrm{O}_{3}$ requires $\left.\mathrm{C}, 76.1 ; \mathrm{H}, 6.8 ; \mathrm{N}, 5.3 \%\right)$; 1-(2,2-diphenyl-2-hydroxy-1-phenyl)ethyl-4,6-diphenylpyrimidin$2(1 \mathrm{H})$-one $(20 \mathrm{~g})(20 \%)$, prisms from $95 \%$ EtOH, m.p. $229^{\circ} \mathrm{C}$ (decomp.) (Found: C, 82.7; H, 5.4; N, 5.3. $\mathrm{C}_{36} \mathrm{H}_{28} \mathrm{~N}_{2} \mathrm{O}_{2}$ requires $\mathrm{C}, 83.0 ; \mathrm{H}, 5.4 ; \mathrm{N}, 5.4 \%)$; 1-[2-hydroxy-2-(4-methylphenyl)-1-phenyl]ethyl-4,6-diphenylpyrimidin-2(1H)one ( 20 h ) ( $40 \%$ ) (isolated by prep. t.l.c.), prisms from $95 \%$ $\mathrm{EtOH}, \mathrm{m} . \mathrm{p} .230^{\circ} \mathrm{C}$ (decomp.) (Found: C, $81.6 ; \mathrm{H}, 5.7$; N, 6.2. $\mathrm{C}_{31} \mathrm{H}_{26} \mathrm{~N}_{2} \mathrm{O}_{2}$ requires C, $\left.81.2 ; \mathrm{H}, 5.7 ; \mathrm{N}, 6.1 \%\right)$; $1-(\alpha-$ methylbenzyl)-4,6-diphenylpyrimidin- $2(1 \mathrm{H}$ )-one (20i) ( $40 \%$ ),
plates from toluene, m.p. $170-170.5^{\circ} \mathrm{C}$ (Found: C, 81.7; $\mathrm{H}, 5.5 ; \mathrm{N}, 7.9 . \quad \mathrm{C}_{24} \mathrm{H}_{20} \mathrm{~N}_{2} \mathrm{O}$ requires $\mathrm{C}, 81.8 ; \mathrm{H}, 5.7 ; \mathrm{N}$, $7.9 \%$ ) ; 1-( $\alpha$-ethylbenzyl)-4,6-diphenylpyrimidin-2(1H)-one $(20 \mathrm{j})(33 \%)$ (isolated by prep. t.l.c.), prisms from $95 \%$ $\mathrm{EtOH}, \mathrm{m} . \mathrm{p} .218{ }^{\circ} \mathrm{C}$ (decomp.) (Found: C, 81.8; H, 6.3; N, 7.6. $\mathrm{C}_{25} \mathrm{H}_{22} \mathrm{~N}_{2} \mathrm{O}$ requires $\left.\mathrm{C}, 81.9 ; \mathrm{H}, 6.0 ; \mathrm{N}, 7.6 \%\right)$ : $1-$ ( $\alpha$-deuteriobenzyl)-4,6-di-p-tolylpyrimidin- $2(1 \mathrm{H}$ )-one (21b) $\left(85 \%\right.$ ), needles from $95 \%$ EtOH, m.p. $207-210^{\circ} \mathrm{C}$ (Found: $\mathrm{C}, 81.5 ; \mathrm{N}, 7.6 . \quad \mathrm{C}_{25} \mathrm{H}_{21} \mathrm{DN}_{2} \mathrm{O}$ requires $\mathrm{C}, 81.7 ; \mathrm{N}, 7.6 \%$ ); 1-( $\alpha$-methylbenzyl)-4,6-di-p-tolylpyrimidin-2(1H)-one (21i) ( $95 \%$ ), prisms from $95 \% \mathrm{EtOH}$, m.p. $186.5-187^{\circ} \mathrm{C}$ (Found: $\mathrm{C}, 81.9 ; \mathrm{H}, 6.5 ; \mathrm{N}, 7.3 . \quad \mathrm{C}_{26} \mathrm{H}_{24} \mathrm{~N}_{2} \mathrm{O}$ requires $\mathrm{C}, 81.7 ; \mathrm{H}$, 6.3 ; $\mathrm{N}, 7.3 \%$ ).

4-n-Butyl-3,4-dihydro-4,6-di-p-tolylpyrimidin-2(1H)-one
(30).-To 1 -methyl-4, 6 -di- $p$-tolylpyrimidin- $2(1 \mathrm{H})$-one (25) ( $1.0 \mathrm{~g}, 3.4 \mathrm{mmol}$ ) in dry THF ( 20 ml ) at $0^{\circ} \mathrm{C}$, under $\mathrm{N}_{2}, \mathrm{n}$ -butyl-lithium ( $0.22 \mathrm{~g}, 3.4 \mathrm{mmol}$ ) was added. Stirring was continued for 1 h at $0^{\circ} \mathrm{C}$ and for a further 0.5 h at $20^{\circ} \mathrm{C}$. Water ( 1 ml ) was added and the solvent removed at $30-40$ ${ }^{\circ} \mathrm{C} / 20 \mathrm{mmHg}$. The residue in EtOAc ( 50 ml ) was washed with water $(2 \times 25 \mathrm{ml})$, dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, and evaporated at $30-40{ }^{\circ} \mathrm{C} / 20 \mathrm{mmHg}$. Prep. t.l.c. [EtOAc-light petroleum (b.p. $60-80^{\circ} \mathrm{C}$ ) ( $65: 35$ )] gave the title compound ( 0.5 g , $42 \%$ ) as needles ( EtOH ), m.p. $147-147.5^{\circ} \mathrm{C}$ (Found: C, 79.1; $\mathrm{H}, 8.3 ; \mathrm{N}, 8.1 . \quad \mathrm{C}_{23} \mathrm{H}_{28} \mathrm{~N}_{2} \mathrm{O}$ requires $\mathrm{C}, 79.3 ; \mathrm{H}$, $8.0 ; \mathrm{N}, 8.0 \%) ; \nu_{\text {max. }}\left(\mathrm{CHBr}_{3}\right) 3200(\mathrm{~N}-\mathrm{H})$ and $1650 \mathrm{~cm}^{-1}$ $(\mathrm{C}=\mathrm{O}) ; \delta\left(\mathrm{CDCl}_{3}\right) 7.3-7.0(8 \mathrm{H}, \mathrm{m}), 5.74(1 \mathrm{H}, \mathrm{d}, J 2$ $\mathrm{Hz}), 4.87(1 \mathrm{H}, \mathrm{d}, J 5 \mathrm{~Hz}), 2.8(3 \mathrm{H}, \mathrm{s}), 2.3(3 \mathrm{H}, \mathrm{s}), 2.28(3$ $\mathrm{H}, \mathrm{s}), 1.82(2 \mathrm{H}, \mathrm{m}), 1.3(2 \mathrm{H}, \mathrm{m})$, and $0.84(3 \mathrm{H}, \mathrm{t}, J 7$ Hz ).

3,6-Dihydro-1,3-dimethyl-6-(1,2-dihydro-2-oxo-4,6-di-p-tolylpyrimiainylmethyl)-4,6-di-p-tolylpyrimidin- $2(1 \mathrm{H})$ one
(28).-LTMP ( 4.5 mmol ) was prepared in situ from n-butyllithium ( $2.86 \mathrm{ml}, 4.5 \mathrm{mmol}$ of 1.0 m ) and 2,2,6,6-tetramethylpiperidine ( $0.63 \mathrm{~g}, 4.5 \mathrm{mmol}$ ), under $\mathrm{N}_{2}$ at $-20^{\circ} \mathrm{C}$. Dry THF ( 6 ml ) was added, the whole cooled to $-76{ }^{\circ} \mathrm{C}$, and 1 -methyl-4,6-di-p-tolylpyrimidin-2( $1 H$ )-one ( $1.0 \mathrm{~g}, 3.5 \mathrm{mmol}$ ) in dry THF ( 25 ml ) added. After 45 min , MeI $(0.6 \mathrm{~g}, 4.9$ mmol ) was added, stirring continued for 1 h at $-76{ }^{\circ} \mathrm{C}$, and for a further 12 h at $20^{\circ} \mathrm{C}$. Water ( 1 ml ) was then added, the solvent removed at $40-50^{\circ} \mathrm{C} / 20 \mathrm{mmHg}$, and the residue taken up in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(60 \mathrm{ml})$, washed with $\mathrm{H}_{2} \mathrm{O}(30 \mathrm{ml})$, and dried (anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$ ). The $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ solution was evaporated at $40-50^{\circ} \mathrm{C} / 20 \mathrm{mmHg}$. Prep. t.l.c. [EtOAclight petroleum (b.p. $60-80{ }^{\circ} \mathrm{C}$ ) (55:45)] gave the title compound ( $0.11 \mathrm{~g}, 10 \%$ ) as prisms, m.p. $286.5-289{ }^{\circ} \mathrm{C}$ (EtOH) (Found: C, 78.8; H, 6.5; N, 9.2. $\mathrm{C}_{39} \mathrm{H}_{38} \mathrm{~N}_{4} \mathrm{O}_{2}$ requires $\mathrm{C}, 78.8 ; \mathrm{H}, 6.4 ; \mathrm{N}, 9.4 \%)$; $v_{\text {max. }}\left(\mathrm{CHBr}_{3}\right) 1650$ $\mathrm{cm}^{-1}(\mathrm{C}=\mathrm{O}) ; \delta\left(\mathrm{CDCl}_{3}\right) 7.92-7.00(16 \mathrm{H}, \mathrm{m}), 6.56(1 \mathrm{H}, \mathrm{s})$, $5.36(1 \mathrm{H}, \mathrm{d}, J 12 \mathrm{~Hz}), 4.75(1 \mathrm{H}, \mathrm{d}, J 12 \mathrm{~Hz}), 4.54(1 \mathrm{H}, \mathrm{s})$, $2.5(3 \mathrm{H}, \mathrm{s}), 2.31(3 \mathrm{H}, \mathrm{s}), 2.28(3 \mathrm{H}, \mathrm{s}), 2.22(3 \mathrm{H}, \mathrm{s}), 2.19$ $(3 \mathrm{H}, \mathrm{s})$, and $1.68(3 \mathrm{H}, \mathrm{s}) ; m / e 594.29\left(M^{+}, 0.12\right), 305.16$ (100), 261.15 ( 0.81 ), and 132.08 (12.48).

3-Ethyl-3,6-dihydro-1-methyl-2-oxo-6-(1,2-dihydro-2-oxo-4,6-di-p-tolylpyrimidinylmethyl)-4,6-di-p-tolylpyrimidin-
$2(1 \mathrm{H})$-one (29).-1-Methyl-4,6-di-p-tolylpyrimidin-2(1H)one ( $1.0 \mathrm{~g}, 3.5 \mathrm{mmol}$ ) in THF ( 25 ml ) was added to a solution [THF ( 10 ml )] of LTMP ( 4.5 mmol ) (prepared as above) at $-76{ }^{\circ} \mathrm{C}$. After 45 min ., EtI ( $0.75 \mathrm{~g}, 4.8 \mathrm{mmol}$ ) was added, stirring continued for 1 h at $-76^{\circ} \mathrm{C}$ and for a further 12 h at $20^{\circ} \mathrm{C}$. Work-up of the reaction was as above. Prep. t.l.c. separation [EtOAc-light petroleum (b.p. 60-80 $\left.\left.{ }^{\circ} \mathrm{C}\right)(55: 45)\right]$ gave the title compound $(0.163 \mathrm{~g}, 16 \%)$ as yellow prisms, m.p. $250-252{ }^{\circ} \mathrm{C}\left(\mathrm{PhCH}_{3}\right)$ (Found: C, 78.8; H, $6.5 ; \mathrm{N}, 9.2 . \mathrm{C}_{40} \mathrm{H}_{40} \mathrm{~N}_{4} \mathrm{O}_{2}$ requires $\mathrm{C}, 78.9 ; \mathrm{H}, 6.6 ; \mathrm{N}$,
$9.2 \%)$; $\nu_{\text {max. }}\left(\mathrm{CHBr}_{3}\right) 1650 \mathrm{~cm}^{-1}(\mathrm{C}=\mathrm{O}) ; \delta\left(\mathrm{CDCl}_{3}\right) 8.0-7.0$ $(16 \mathrm{H}, \mathrm{m}), 6.64(1 \mathrm{H}, \mathrm{s}), 5.34(1 \mathrm{H}, \mathrm{d}, J 12 \mathrm{~Hz}), 4.76(1 \mathrm{H}, \mathrm{d}$, $J 12 \mathrm{~Hz}), 4.56(1 \mathrm{H}, \mathrm{s}), 3.90(1 \mathrm{H}$, sextet), $3.10(1 \mathrm{H}$, sextet), $2.36(6 \mathrm{H}, \mathrm{s}), 2.26(3 \mathrm{H}, \mathrm{s}), 2.24(3 \mathrm{H}, \mathrm{s}), 1.70(3 \mathrm{H}, \mathrm{s})$, and $0.78(3 \mathrm{H}, \mathrm{t}) ; m / e 608.31\left(M^{+}, 0.05\right), 319.17$ (100), 305.16 (2.22), and 132.08 (2.82).

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