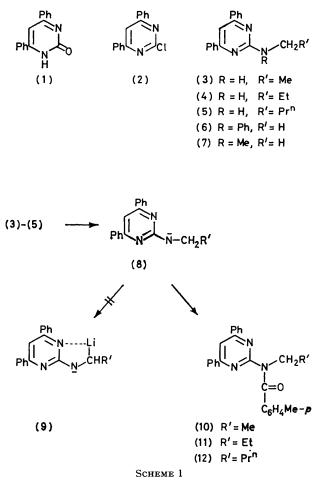
Metallation Studies with Pyrimidines

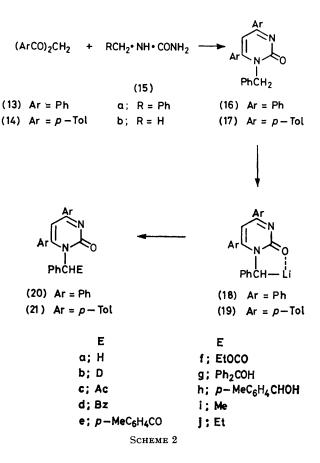
By Alan R. Katrizky,* Hector J. Salgado, Amornsri Chermprapai, and Narayan K. Ponkshe, School of Chemical Sciences, University of East Anglia, Norwich, NR4 7TJ, England and Department of Chemistry, University of Florida, Gainesville, Florida 32611, U.S.A.

2-Alkylamino-4,6-diphenylpyrimidines are acylated only at nitrogen after treatment with lithium di-isopropylamide (LDA). 4,6-Diaryl-1-benzylpyrimidin-2(1*H*)-ones can be acylated and alkylated at the α -CH₂ group. 1-Methyl-4,6-di-*p*-tolylpyrimidin-2(1*H*)-one forms a carbanion which undergoes dimerisation.

As part of a larger study of heterocycle stabilised carbanions,¹ we have investigated pyrimidines: (a) 2alkylamino-4,6-diphenylpyrimidines to attempt to generate carbanions of type (9), and (b) 1-substituted 4,6diphenylpyrimidin-2(1*H*)-ones to examine carbanions of type (18).



2-Alkylaminopyrimidines.—4,6-Diphenylpyrimidin-2(1*H*)-one (1) is converted by phosphoryl chloride into 2chloro-4,6-diphenylpyrimidine (2).² This reacts readily with primary and secondary amines (ethylamine, npropylamine, n-butylamine, and *N*-methylaniline) to give the corresponding 2-alkylaminopyrimidines (3)—(6) (Table 1). The analogous 2-dimethylamino-derivative



(7) was obtained by ring synthesis from dibenzoylmethane and NN-dimethylguanidine. Structures of all the compounds were confirmed by their spectral data (Table 2).

Treatment of the 2-alkylamino-4,6-diphenylpyrimidines (3)—(5) with lithium di-isopropylamide (LDA) gave the mono-anion (8), but not the dianion (9): subsequent treatment with p-toluoyl chloride yielded amides (10)—(12) (Table 3). Structures were confirmed by their spectral data (Table 4): in particular the low v(C=O)excludes the possibility that the acyl group is attached to a cyclic nitrogen atom. Use of excess LDA mainly gave the known 3 NN-di-isopropyl-p-toluamide.

4,6-Diaryl-1-benzylpyrimidin-2(1H)-ones.—The diaroylmethanes (13) and (14) reacted with benzylurea (15a) to give the 4,6-diaryl-1-benzylpyrimidin-2(1H)-

TABLE 1

Preparation of 2-alkylamino-4,6-diphenylpyrimidines (3)-(7)

			-		5	· 1 J	1 2	· · ·	· · ·				
				Yield			Found (%)			Required (%)			
No.	R	R'	M.p. (°C)	(%)	Crystal form	Cryst. solvent	́с	н	Ň	΄ C	н	N	
(3)	Н	Me	6668	49	Needles	EtOH	78.2	6.1	15.1	78.5	6.2	15.3	
(4)	Н	Et	96 - 99	51	Prisms	EtOH	78.7	6.7	14.5	78.9	6.6	14.5	
(5)	H	Pr ⁿ	58 - 61	58	Plates	EtOH	78.8	6.8	13.7	79.2	7.0	13.8	
(6)	Ph	H	147 - 147.5	91	Fine needles	95% EtOH	81.8	5.8	12.4	81.9	5.7	12.5	
(7)	Me	н	131.5 - 132.5	30	Prisms	EtOH	78.5	6.0	15.2	78.5	6.2	15.3	

TABLE 2

I.r. ^a and ¹H n.m.r. (δ , 60 MHz) ^b spectral data of 2-alkylamino-4,6-diphenylpyrimidines (3)-(7)

			I.r.			N.m.r.			
No.	R	R′	(cm^{-1})	N R	CH ₂	F	R′		
(3)	Н	Me	3 280	5.35 (1 H, bm)	3.65 (2 H, qn)	1.33 (3 H, t)		7.8 (11 H, m)	
(4)	н	Et	3 250	5.4 (1 H, bm)	3.5 (2 H, q)	1.5 (2 H, m)	1.0 (3 H, t)	7.8 (11 H, m)	
(5)	Н	Pr ⁿ	3 280	5.9 (1 H, t)	centred at 3.35 (2 H, bm)	0.9 (3 H, d)	1.4 (4 H, m)	7.8 (11 H, m)	
(6)	\mathbf{Ph}	Н			3.68 (3 H, m)			7.12-7.68 (12 H, m) 7.9-8.20 (4 H, m)	
(7)	Me	Н			3.34 (6 H, s)			7.36-7.66 (7 H, m) 8.17-8.2 (4 H m)	

" In CHBr₃. In CDCl₃. s = singlet, bm = broad multiplet, t = triplet, q = quartet, qn = quintet, m = multiplet.

TABLE 3

Preparation of N-alkyl-N-(4,6-diphenylpyrimidinyl)-p-toluamides (10)-(12)

			Yield				Found (%)	Required (%)				
No.	R'	M.p. (°C)		Cryst. form	Cryst. solvent	C	Н	N	Ċ	Н	N		
(10)	Me	128-132	60	Needles	Ether-light petroleum (b.p. 60-80 °C)	78.9	5.6	10.6	79.4	5.9	10.7		
(11) (12)	Et Pr ⁿ	$135-138 \\ 91-94$	$\frac{55}{37}$	Prisms Needles	ÈtÔH 95% EtOH	$79.4 \\ 79.5$	$\begin{array}{c} 6.4 \\ 6.5 \end{array}$	$\begin{array}{c} 10.3 \\ 9.9 \end{array}$	$79.6 \\ 79.8$	$\begin{array}{c} 6.2 \\ 6.5 \end{array}$	$\begin{array}{c} 10.3 \\ 10.0 \end{array}$		

TABLE 4

I.r. a and ¹H n.m.r. (δ, 60 MHz) b spectral data of N-alkyl-N-(4,6-diphenylpyrimidinyl)-p-toluamides (10)-(12)

		I.r.	N.m.r.										
No.	R′	(cm ⁻¹) >=0	<i>p</i> -Toluoyl	N-CH2	R'		Aromatic						
(10)	Me	1 655	2.27 (3 H, s)	4.42 (2 H, m)	1.4 (3 H, t)		7.4 (15 H, m)						
(11)	Et	1 650	(3 H, s) 2.25 (3 H, s)	(2 H, H) 4.35 (2 H, t)	(3 H, t) centred at 1.9 (2 H, m)	1.05 (3 H, t)	(15 H, m) 7.5 (15 H, m)						
(12)	Pr ⁿ	1 650	2.28 (3 H, s)	4.37 (2 H, t)	(2 H, H) 1.0 (3 H, t)	1.65 (4 H, m)	7.5 (15 H, m)						
	ª In C	CHBr₃. ⁰In	CDCl ₃ . s =	singlet, m	= multiplet, t	= triplet.							

ones (16) and (17) (cf. ref. 4). We have shown previously that 1-benzyl-4,6-diphenyl-2(1*H*)-pyridone (22) is converted by LDA into the lithio-derivative (23) which reacts with electrophiles to form α -substituted products (24). We now find that lithio-derivatives (18) and (19) can be formed similarly; they show an intense blue colouration.

The diphenyl carbanions (18) and (19) react with D_2O to form the deuteriated pyrimidinones (20b) and (21b). Organolithium (18) adds to a variety of electrophiles to

form products (20c-j); methyl and ethyl iodide gave the alkylated derivatives (20i) and (20j); ethyl chloroformate the ester (20f); acetyl, benzoyl, and *p*-toluoyl chloride the ketones (20c), (20d), and (20e); *p*-tolualdehyde and benzophenone the hydroxy-derivatives (20h) and (20g). Organolithium (19) reacted with methyl iodide to give the alkylated derivative (21i). All the compounds were characterised by their spectral data (Table 5).

1-Methyl-4,6-di-p-tolylpyrimidin-2(1H)-one.—Reaction

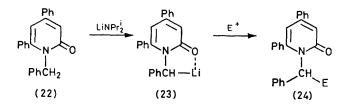
 TABLE 5

 I.r. a and ¹H n.m.r. (8, 60 MHz) b spectral data of 4,6-diaryl-1-benzylpyrimidin-2-ones (20) and (21)

		I.	r. (cm ⁻¹)		¹ H N.m.r.							
No.	Alpha substituent	' Pyridone C=O			5-H (s)		R	۸	Aromatic (m)			
(20a)	Н	1 645	•••		6.70	5.2 (s)			7.52—7.05 (15 H)			
(20b)	D	1 650			6.70	5.2 (s) °			7.05—7.52 (15 H)			
(20c)	MeCO	1 640		1680	6.42	d		2.3 (s)	6.80—7.82 (16 H)			
(20d)	PhCO	1650		1 700	6.80	6.65 (s)			7.2—8.3 (20 H)			
(20e)	p-MeC ₆ H ₄ CO	1650		1 690	6.72	d		2.28 (s)	7.0—8.2 (20 H)			
(20f)	EtOCO	1650		1 740	6.40	6.30 (s)	4.0 (m)	1.1 (t)	6.98—7.9 (15 H)			
(20g)	Ph _s COH	1 650	$3\ 250$		6.16	6.0 (s)			6.74-7.3 (25 H)			
(20h)	p-MeC ₆ H₄CHOH	1 650	3 300		6.50	5.86 (d) e	5.45 (d) e	2.24 (s)	7.0—8.0 (14 H)			
(20i)	Me	1 650			6.90	5.5 (q)	. ,	1.6 (d)	7.2—7.8 (15 H)			
(20i)	Et	1655			6.50	5.9 (m)	1.2 (m)	0.7 (t)	6.9—7.9 (15 H)			
$(21a)^{f}$	H	1 650			6.60	5.2 (s)	• • •	()	7.9—7.1 (13 H)			
(21b) f	Ď	1 655			6.60	5.15 (s) °			7.08.0 (13 H)			
$(21i)^{f}$	Me	1 650			6.50	5.4 (q)		1.75 (d)	7.85—7.05 (13 H)			

^a In CHBr₃. ^b In CDCl₃. d = doublet, q = quartet, s = singlet, t = triplet, m = multiplet. ^c Integrates for 1 H. ^d Overlapped in the aromatic region. ^e CH, doublet (J 10 Hz). ^f In addition two singlets (3 H each) are shown at 2.3 and 2.4 p.p.m.

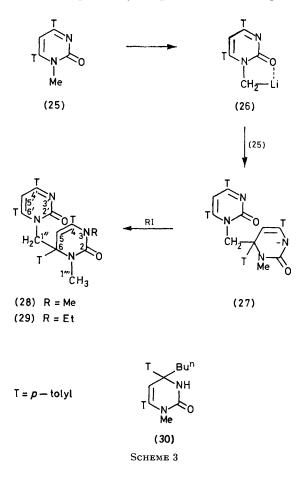
of N-methylurea (15b) with (14) yielded 1-methylpyrimidinone (25) which formed a deep red colouration with LDA; subsequent addition of electrophiles then merely gave recovered (25). Attempted use of n-butyllithium as base afforded the addition product (30), which shows $\nu(NH)$ 3 200 cm⁻¹ and $\nu(C=O)$ 1 650 (broad).



In the ¹H n.m.r., the n-butyl group signals occur at 0.84 (3 H, t), 1.3 (4 H, m), and 1.82 p.p.m. (2 H, m). The *C*-methyls resonate as two 3-H singlets near to 2.8 p.p.m. The heterocyclic ring olefinic hydrogens form a 2 Hz split doublet at 4.87 p.p.m. coupled (J 2 Hz) with the N-H proton (broad doublet at 5.74 p.p.m.). These spectral data appear to support a 1,2-addition to C=N to give (30) rather than 1,4-addition to C=C-C=N to give an isomer, but we have not rigorously excluded the isomeric structure. Sodium borohydride reduces pyrimidin-2-ones to dihydro- and tetrahydro-derivatives,⁵ and this provides an analogy for the formation of (30).

However, reaction of (25) with 2,2,6,6-tetramethylpiperidyl-lithium (LTMP) followed by methyl iodide formed (28) by a 1,6-addition of the lithio-derivative (26) to a second molecule of (25) to give anion (27). Similarly addition of ethyl iodide produced (29).

The structures (28) and (29) are based on spectral evidence: both (28) and (29) showed v(C=O) (pyrimidinone) at 1 650 cm⁻¹ (broad). In the ¹H n.m.r. spectra, the ring proton of the undisturbed pyrimidinone ring in (28) gave a singlet at δ 6.56 (5'-H) and at δ 6.62 in (29) (5'-H). The corresponding signal in the pyrimidinone (25) was at δ 6.64. In the modified pyrimidinone ring of (28) and (29), the 5-H signal shifted upfield, resonating as a singlet at δ 4.54 and 4.56 respectively. The prochiral protons of the bridging methylene group were magnetically non-equivalent and gave an AB system in both compounds. In (28) the A-proton appeared at δ 4.75 and the B-proton at δ 5.34 with J_{gem} 12 Hz. Similarly in (29) the A-proton appeared at δ 4.76 and the B-proton at δ 5.41 with J_{gem} 14 Hz. In (28) and (29) the N-methyl group (1"' protons) gave a singlet at δ 1.70; in (28) the incorporated methyl gave a singlet at δ 2.5 and in (29), the diastereotopic methylene protons of the incorporated



ethyl formed two sextets at δ 3.10 and δ 3.90 with $J_{\rm vic}$ 7 Hz and $J_{\rm gem}$ 14 Hz. The methyl group displayed a triplet at δ 0.78 p.p.m. The aromatic protons in both (28) and (29) displayed a multiplet in the range δ 8.0—7.0.

In the ¹³C n.m.r. spectra, with off-resonance C-H information, for both (28) and (29), the undisturbed pyrimidinone carbonyl carbon gave a singlet at δ 169.19 and at 169.23 respectively [cf. 169.31 in (25)] whilst the modified pyrimidinone carbonyl resonated at δ 160.27 and 160.12. The aromatic carbons displayed a series of

HA-100 (100 MHz) n.m.r. spectrometers, ¹³C n.m.r. spectra at 25.05 MHz on a Jeol FX-100 Fourier transform spectro-photometer, and high-resolution mass spectra on an AEI MS-9 spectrometer.

4,6-Diphenylpyrimidin-2-one (1).—Dibenzoylmethane (5 g, 22.3 mmol), urea (2.0 g, 33.4 mmol), toluene-*p*-sulphonic acid (5.75 g, 33.4 mmol), and glacial HOAc (15 ml) were heated under reflux for 48 h and neutralised with aqueous NaOH (12%). The product separated; it was collected, washed with water, dried, and crystallised (EtOH) (m.p. 237—239 °C) (4.20 g, 15%) [lit.,^{2b} m.p. 237—239 °C].

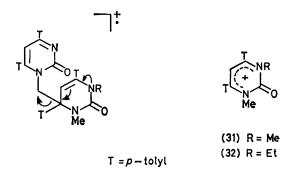
TABLE 6

¹³C N.m.r. spectra ^a of 1,3-disubstituted 1,2,3,6-tetrahydro-6-(1,2-dihydro-2-oxo-4,6-di-p-tolylpyrimidinylmethyl)-4,6-di-p-tolylpyrimidin-2(1H)-ones (28) and (29)

	Cart	onvl	Aromatic	Olefinic region			Aliphatic region								
Compd. no.	region (s)		region Unassigned multiplet	5-C 5'-C (d) (d)			6C (s)	1′′′-C 1′′′′-C (t) (q)			· · .	-СН ₃ q)	3-N-R		
$(28) \\ (29)$		$\begin{array}{c} 160.27\\ 160.12 \end{array}$	$159.68 - 126.64 \\ 159.39 - 126.55$				$\begin{array}{c} 67.15 \\ 67.73 \end{array}$	48.53 48.04	$\begin{array}{c} 30.35\\ 30.26\end{array}$	$\begin{array}{c} 21.48\\ 21.50 \end{array}$	$\begin{array}{c} 21.38\\ 21.39 \end{array}$	$\begin{array}{c} 21.28\\21.19\end{array}$	$\begin{array}{c} 20.91 \\ 20.90 \end{array}$	33.08 (q) 39.27 (t)	14.96 (q)
	^{a} In CDCl ₃ with SiMe ₄ as internal reference.														

lines in the range 159.68—126.55 (Table 6). The 5 and 5' ring carbons gave doublets at δ 105.30 and δ 103.60 in (28) and at δ 105.20 and δ 103.84 in (29) [cf. 102.44 in (25)]. In the aliphatic region, the bridging methylene carbon gave a triplet at δ_{av} . 48, the 6-quaternary carbon a singlet at δ_{av} . 67, and the N-methyl, a quartet at δ_{av} . 30 (Table 6). The inserted methyl in (28) gave a quartet at δ 33.08 whilst in (29), the inserted ethyl gave a triplet at δ 39.27 and a quartet at 14.96. The C-methyl group gave 4 quartets in the range δ 21.50—20.90 (Table 6).

High-resolution mass spectroscopy showed the molecular ion peak at m/e 594.29 (0.12%) for (28) and at



608.31 (0.05%) for (29). Both (28) and (29) underwent a retrosynthetic expulsion of the oxopyrimidinylmethyl fragment to give the observed base peaks at m/e 305.16 (100%) (31) and 319.17 (100%) (32) respectively.

EXPERIMENTAL

M.p.s were measured on a Reichert hot-stage melting point apparatus and are uncorrected. I.r. spectra were recorded in CHBr₃ on a Perkin-Elmer 297 spectrophotometer, ¹H n.m.r. spectra on Perkin-Elmer R12 (60 MHz) and Varian

2-Chloro-4,6-diphenylpyrimidine (2) (47%), had m.p. 113-113.5 °C [lit.,^{2a} m.p. 115-116 °C].

General Procedure for the Preparation of 2-Amino-4,6diphenylpyrimidines (3)—(6).—2-Chloro-4,6-diphenylpyrimidine (2) and the appropriate amine were heated under reflux in absolute EtOH for 2 h. [For the preparation of (3), dry EtNH₂ gas was bubbled through a solution of (2) in absolute EtOH and for (6), the reactants were heated in the absence of absolute EtOH]. Evaporation (100 °C/15 mmHg) gave the crude product, which was washed with water and crystallised (see Table 1).

2-Dimethylamino-4,6-diphenylpyrimidine (7).—Dibenzoylmethane (1 g, 4.46 mmol), NN-dimethylguanidine, HCl (1 g, 8.16 mmol), and K_2CO_3 (0.6 g, 4.3 mmol) were heated under reflux in EtOH (10 ml) for 2 h then extracted with CH_2Cl_2 (50 ml) and the extracts washed with H_2O (20 ml). The dried (MgSO₄) extracts on evaporation (100 °C/15 mmHg) furnished the product, which was crystallised (see Table 1).

General Procedure for the Preparation of Amides (10)-(12). -To LDA (1 mmol) in dry THF (5 ml) [prepared by adding dropwise n-butyl-lithium in hexane (1 mmol) to di-isopropylamine (1 mmol) at 0 to -5 °C under N₂] cooled to 0 to -5 °C was added 2-alkylaminopyrimidine (3)-(5) (0.5 mmol) in dry THF (2 ml). After 0.5 h at 0 to -5 °C, p-toluoyl chloride (1 mmol) in dry THF (2 ml) was added. Stirring was continued for a further 4 h at 20 °C, water (5 ml) was then added and the solution extracted with CH₂Cl₂ (30 ml); the extracts were washed with aqueous NaHCO₃ (10%, 10 ml) followed by H₂O (10 ml). The dried (Na₂SO₄) extracts on evaporation (100 °C/15 mmHg) gave the crude product as a yellowish oil which solidified slowly and crystallised from the appropriate solvent (Table 3). Purification of the motherliquor afforded NN-di-isopropyl-p-toluamide as the byproduct, m.p. 84 °C [lit.,3 m.p. 85-86 °C] as colourless plates [toluene-light petroleum (b.p. 40-60 °C)].

l-Benzyl-4,6-diphenylpyrimidin-2(1H)-one (20a).—Dibenzoylmethane (5.0 g, 20 mmol), benzylurea (4.8 g, 32 mmol), and toluene-p-sulphonic acid (7.6 g, 40 mmol) in glacial HOAc (8 ml) were heated at reflux for 36 h. Cooling and treatment with aqueous (50%) EtOH gave the pyrimidinone which crystallised from 95% EtOH as prisms (5.0 g, 68%), m.p. 164—165 °C (Found: C, 81.3; H, 5.4; N, 8.3. C_{23} -H₁₈N₂O requires C, 81.6; H, 5.3; N, 8.3%).

1-Benzyl-4,6-di-p-tolylpyrimidin-2(1H)-one (21a).—Di-ptoluoylmethane (1.0 g, 4 mmol), benzylurea (0.95 g, 6 mmol), and toluene-p-sulphonic acid (1.4 g, 7 mmol), in HOAc (1 ml) were heated at reflux for 36 h. Work-up as described for (20a) gave the pyrimidinone (21a) (0.9 g, 62%) as needles from 95% EtOH, m.p. 209—210 °C (Found: C, 81.9; H, 6.1; N, 7.6. $C_{25}H_{22}N_2O$ requires C, 82.0; H, 6.0; N, 7.6%). 1-Methyl-4,6-di-p-tolylpyrimidin-2(1H)-one (25).—Di-p-

toluoylmethane (5.0 g, 19 mmol), methylurea (3.1 g, 40 mmol), and toluene-*p*-sulphonic acid (7.2 g, 37 mmol) in HOAc (9 ml) were heated at reflux for 36 h. Work-up as described for (20a) gave the *pyrimidinone* (25) (3.5 g, 61%) as needles from 95% EtOH, m.p. 197—199 °C (Found: C, 78.7; H, 6.3; N, 9.6. $C_{19}H_{18}N_2O$ requires C, 78.6; H, 6.2; N, 9.6%); ν_{max} (CHBr₃) 1 645 cm⁻¹ (C=O); δ (CDCl₃) 7.98—7.22 (8 H, m), 6.70 (1 H, s), 2.40 (3 H, s), 2.39 (3 H, s), and 3.40 (3 H, s).

General Procedure for the Lithiation and Alkylation of 4,6-Diaryl-1-benzylpyrimidin-2(1H)-ones.—LDA (3.0 mmol) was prepared by adding dropwise di-isopropylamine (0.3 g, 3.0 mmol) to n-butyl-lithium in hexane (3.1 ml, 3.0 mmol of 0.96M) at -20 °C under N₂. Stirring of the mixture was continued until it became cloudy when dry THF (6 ml) added; the whole was then cooled to -76 °C and 4,6-diaryl-1-benzylpyrimidin-2-one (3.0 mmol) in dry THF (20 ml) was added. After 40 min at -76 °C, the electrophile (3.0 mmol) in dry THF (5 ml) was added. Stirring was continued for 1 h at -76 °C and for a further 10 h at 20 °C. Water (1 ml) was then added and the solvent removed at 40-50 °C/20 mmHg. The residue in CH₂Cl₂ (70 ml) was washed with saturated aqueous NaCl (30 ml) and water (30 ml) and then dried (Na₂SO₄), and evaporated at 40–50 $^{\circ}\text{C}/20$ mmHg. The product was separated by crystallisation [CHCl₃-light petroleum (b.p. 60-80 °C)], or prep. t.l.c. and further recrystallised from the appropriate solvent.

The following compounds were prepared according to the general procedure: $1-(\alpha - deuteriobenzyl)-4, 6-diphenylpyrimi$ din-2(1H)-one (20b) (90%), prisms from EtOH, m.p. 160-161 °C (Found: C, 81.5; N, 8.2. C₂₃H₁₇DN₂O requires C, 81.4; N, 8.2%); $1-(\alpha$ -acetylbenzyl)-4,6-diphenylpyrimidin-2(1H)-one (20c) (48%), plates from EtOH, m.p. 184-185 °C (Found: C, 78.7; H, 5.7; N, 7.4. C₂₅H₂₀N₂O₂ requires C, 78.9; H, 5.3; N, 7.4%); 1-(a-benzoylbenzyl)-4,6-diphenylpyrimidin-2(1H)-one (20d) (52%), plates from EtOH, m.p. 168—170 °C (Found: C, 81.4; H, 5.0; N, 6.3. C₃₀H₂₂N₂O₂ requires C, 81.4; H, 5.0; N, 6.3%); 4,6-diphenyl-1-[a-(4toluoyl)benzyl]pyrimidin-2(1H)-one (20e) (54%), plates from EtOH, m.p. 140-142 °C (Found: C, 81.2; H, 5.6; N, 6.0. $C_{31}H_{24}N_2O_2$ requires C, 81.6; H, 5.3; N, 6.1%); 1-(α ethoxycarbonylbenzyl)-4,6-diphenylpyrimidin-2(1H)-one (20f) (35%) (isolated by prep. t.l.c.), prisms from 95% EtOH, m.p. 228-230 °C (Found: C, 76.0; H, 6.8; N, 5.3. C₂₆- $H_{22}N_2O_3$ requires C, 76.1; H, 6.8; N, 5.3%); 1-(2,2diphenyl-2-hydroxy-1-phenyl)ethyl-4,6-diphenylpyrimidin-2(1H)-one (20g) (20%), prisms from 95% EtOH, m.p. 229 °C (decomp.) (Found: C, 82.7; H, 5.4; N, 5.3. $C_{36}H_{28}N_2O_2$ requires C, 83.0; H, 5.4; N, 5.4%); 1-[2-hydroxy-2-(4methylphenyl)-1-phenyl]ethyl-4,6-diphenylpyrimidin-2(1H)one (20h) (40%) (isolated by prep. t.l.c.), prisms from 95%EtOH, m.p. 230 °C (decomp.) (Found: C, 81.6; H, 5.7; N, 6.2. $C_{31}H_{26}N_2O_2$ requires C, 81.2; H, 5.7; N, 6.1%); 1-(α methylbenzyl)-4,6-diphenylpyrimidin-2(1H)-one (20i) (40%), 157

plates from toluene, m.p. 170-170.5 °C (Found: C, 81.7; H, 5.5; N, 7.9. $C_{24}H_{20}N_2O$ requires C, 81.8; H, 5.7; N, $1-(\alpha-ethylbenzyl)-4, 6-diphenylpyrimidin-2(1H)-one$ 7.9%); (20i) (33%) (isolated by prep. t.l.c.), prisms from 95%EtOH, m.p. 218 °C (decomp.) (Found: C, 81.8; H, 6.3; N, 7.6. C₂₅H₂₂N₂O requires C, 81.9; H, 6.0; N, 7.6%): 1- $(\alpha$ -deuteriobenzyl)-4,6-di-p-tolylpyrimidin-2(1H)-one (21b)(85%), needles from 95% EtOH, m.p. 207-210 °C (Found: C, 81.5; N, 7.6. $C_{25}H_{21}DN_2O$ requires C, 81.7; N, 7.6%); $1-(\alpha-methylbenzyl)-4, 6-di-p-tolylpyrimidin-2(1H)-one$ (21i)(95%), prisms from 95% EtOH, m.p. 186.5-187 °C (Found : C, 81.9; H, 6.5; N, 7.3. C₂₆H₂₄N₂O requires C, 81.7; H, 6.3; N, 7.3%).

4-n-Butyl-3,4-dihydro-4,6-di-p-tolylpyrimidin-2(1H)-one (30).—To 1-methyl-4,6-di-p-tolylpyrimidin-2(1H)-one (25) (1.0 g, 3.4 mmol) in dry THF (20 ml) at 0 °C, under N_2 , nbutyl-lithium (0.22 g, 3.4 mmol) was added. Stirring was continued for 1 h at 0 °C and for a further 0.5 h at 20 °C. Water (1 ml) was added and the solvent removed at 30-40°C/20 mmHg. The residue in EtOAc (50 ml) was washed with water (2 imes 25 ml), dried (Na₂SO₄), and evaporated at 30-40 °C/20 mmHg. Prep. t.l.c. [EtOAc-light petroleum (b.p. 60-80 °C) (65:35)] gave the title compound (0.5 g, 42%) as needles (EtOH), m.p. 147-147.5 °C (Found: C, 79.1; H, 8.3; N, 8.1. C₂₃H₂₈N₂O requires C, 79.3; H, 8.0; N, 8.0%); $\nu_{max.}$ (CHBr₃) 3 200 (N⁻H) and 1 650 cm⁻¹ (C=O); δ (CDCl₃) 7.3–7.0 (8 H, m), 5.74 (1 H, d, J 2 Hz), 4.87 (1 H, d, J 5 Hz), 2.8 (3 H, s), 2.3 (3 H, s), 2.28 (3 H, s), 1.82 (2 H, m), 1.3 (2 H, m), and 0.84 (3 H, t, J 7 Hz).

3,6-Dihydro-1,3-dimethyl-6-(1,2-dihydro-2-oxo-4,6-di-ptolylpyrimidinylmethyl)-4,6-di-p-tolylpyrimidin-2(1H)one (28).—LTMP (4.5 mmol) was prepared in situ from n-butyllithium (2.86 ml, 4.5 mmol of 1.0M) and 2,2,6,6-tetramethylpiperidine (0.63 g, 4.5 mmol), under N_2 at -20 °C. Dry THF (6 ml) was added, the whole cooled to -76 °C, and 1methyl-4,6-di-p-tolylpyrimidin-2(1H)-one (1.0 g, 3.5 mmol) in dry THF (25 ml) added. After 45 min, MeI (0.6 g, 4.9 mmol) was added, stirring continued for 1 h at -76 °C, and for a further 12 h at 20 °C. Water (1 ml) was then added, the solvent removed at 40-50 °C/20 mmHg, and the residue taken up in CH_2Cl_2 (60 ml), washed with H_2O (30 ml), and dried (anhydrous Na₂SO₄). The CH₂Cl₂ solution was evaporated at 40-50 °C/20 mmHg. Prep. t.l.c. [EtOAclight petroleum (b.p. 60-80 °C) (55:45)] gave the title compound (0.11 g, 10%) as prisms, m.p. 286.5-289 °C (EtOH) (Found: C, 78.8; H, 6.5; N, 9.2. C₃₉H₃₈N₄O₂ requires C, 78.8; H, 6.4; N, 9.4%); $\nu_{max.}$ (CHBr₃) 1 650 cm⁻¹ (C=O); δ (CDCl₃) 7.92–7.00 (16 H, m), 6.56 (1 H, s), 5.36 (1 H, d, J 12 Hz), 4.75 (1 H, d, J 12 Hz), 4.54 (1 H, s), 2.5 (3 H, s), 2.31 (3 H, s), 2.28 (3 H, s), 2.22 (3 H, s), 2.19 (3 H, s), and 1.68 (3 H, s); m/e 594.29 (M^+ , 0.12), 305.16 (100), 261.15 (0.81), and 132.08 (12.48).

3-Ethyl-3,6-dihydro-1-methyl-2-oxo-6-(1,2-dihydro-2-oxo-4,6-di-p-tolylpyrimidinylmethyl)-4,6-di-p-tolylpyrimidin-2(1H)-one (29).—1-Methyl-4,6-di-p-tolylpyrimidin-2(1H)-one (1.0 g, 3.5 mmol) in THF (25 ml) was added to a solution [THF (10 ml)] of LTMP (4.5 mmol) (prepared as above) at -76 °C. After 45 min., EtI (0.75 g, 4.8 mmol) was added, stirring continued for 1 h at -76 °C and for a further 12 h at 20 °C. Work-up of the reaction was as above. Prep. t.1.c. separation [EtOAc-light petroleum (b.p. 60—80 °C) (55: 45)] gave the *title compound* (0.163 g, 16%) as yellow prisms, m.p. 250—252 °C (PhCH₃) (Found: C, 78.8; H, 6.5; N, 9.2. $C_{40}H_{40}N_4O_2$ requires C, 78.9; H, 6.6; N,

9.2%); v_{max} (CHBr₃) 1 650 cm⁻¹ (C=O); δ (CDCl₃) 8.0-7.0 (16 H, m), 6.64 (1 H, s), 5.34 (1 H, d, J 12 Hz), 4.76 (1 H, d, J 12 Hz), 4.56 (1 H, s), 3.90 (1 H, sextet), 3.10 (1 H, sextet), $2.36\ (6\ H,\ s),\ 2.26\ (3\ H,\ s),\ 2.24\ (3\ H,\ s),\ 1.70\ (3\ H,\ s),\ and$ 0.78 (3 H, t); m/e 608.31 (M^+ , 0.05), 319.17 (100), 305.16 (2.22), and 132.08 (2.82).

We thank the Leverhulme Foundation for a Fellowship (to N. K. P.) and the Consejo Nacional de Ciencia y Tecnologia (Mexico) for a scholarship (to H. J. S. Z.).

[1/188 Received, 9th February, 1981]

REFERENCES

¹ (a) A. R. Katritzky, N. E. Grzeskowiak, T. Siddiqui, C. Jayaram, and S. N. Vassilatos, J. Chem. Res., in the press; (b) A. R. Katritzky, J. Arrowsmith, Zakaria bin Bahari, C. Jayaram, T. R. Katritzky, J. Arrowsmith, Zakaria bin Bahari, C. Jayaram, T. Siddiqui, and S. Vassilatos, J. Chem. Soc., Perkin Trans. 1, 1980, 2851; (c) A. R. Katritzky, N. E. Grzeskowiak, H. J. Salgado, and Zakaria bin Bahari, Tetrahedron Lett., 1980, 21, 4451.
^a (a) CIBA Ltd., Fr. P. 1, 396, 684/1965 [Chem. Abstr., 1965, 63, 99653]; (b) V. P. Mamaev, Biol. Akt. Soedin., Akad Nauk SSSR, 1965, 38, [Chem. Abstr., 1965, 63, 180817].
^a R. E. Ludt, J. S. Griffiths, K. N. McGrath, and C. R. Hauser, J. Org. Chem., 1973, 38, 1668.
⁴ G. E. Hardtmann and F. G. Kathawala, Ger. P. 2,056,406/ 1971 [Chem. Abstr., 17, 75, 63813m].

1971 [Chem. Abstr., 1971, 75, 63813m].
 ⁵ C. Kashima, Y. Yokota, T. Nishio, and A. Katoh. Hetero-

cycles, 1980, 14, 120.